

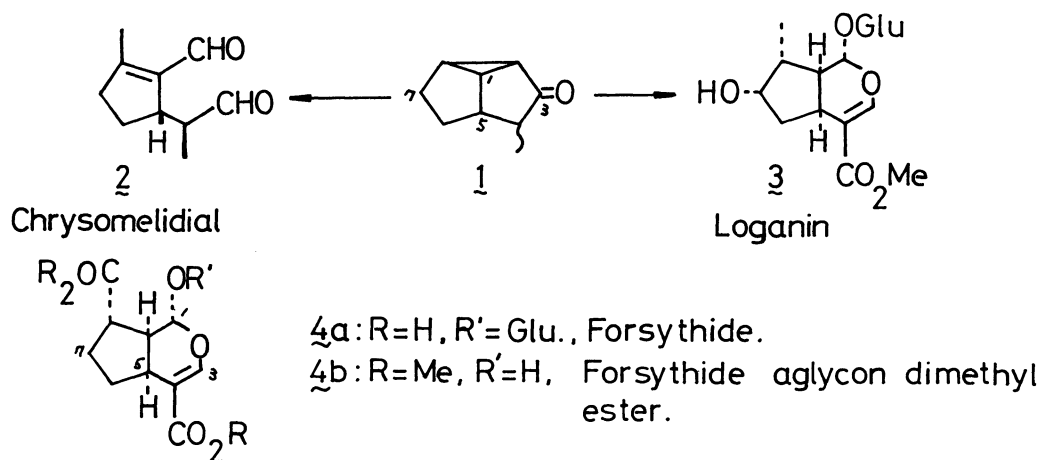
TRICYCLO[3.3.0.0^{2,8}]OCTANONES AS BUILDING BLOCKS IN NATURAL PRODUCTS
SYNTHESIS. -- SYNTHESIS OF (±)-FORSYTHIDE AGLYCON DIMETHYL ESTER --

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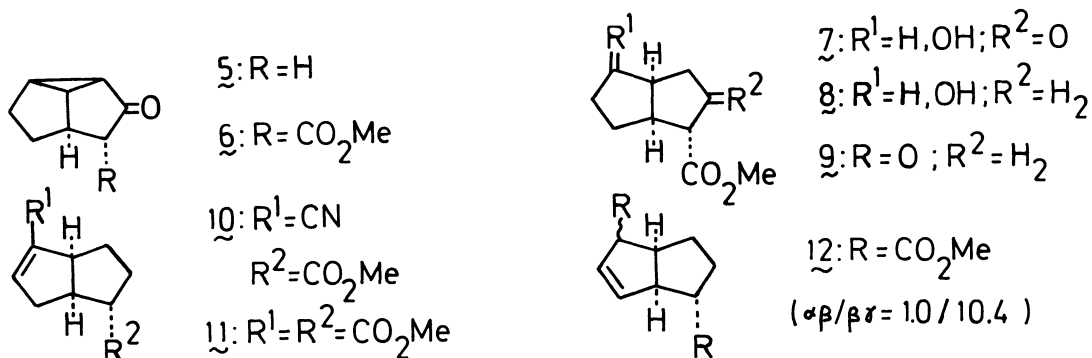
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The stereocontrolled synthesis of (±)-forsythide aglycon dimethyl ester from 4-methoxycarbonyl-tricyclo[3.3.0.0^{2,8}]octan-3-one (6) is described.

The various methodologies have been extended towards the synthesis of cyclopentanoid natural products. We recently identified the potential utility of 4-methyl-tricyclo[3.3.0.0^{2,8}]octan-3-one (1) as a versatile intermediate in the synthesis of chrysolidial¹⁾ (2) and loganin²⁾ (3). In this paper we wish to report the stereocontrolled synthesis of (±)-forsythide aglycon dimethyl ester (4b), starting from another versatile synthon, 4-methoxycarbonyl-tricyclo[3.3.0.0^{2,8}]octan-3-one (6). Forsythide (4a)³⁾ is a naturally occurring iridoid glucoside isolated from fresh leaves of *Forsythia viridissima* Lindl.



The key intermediate 6 was prepared from tricyclo[3.3.0.0^{2,8}]octan-3-one (5)⁴⁾ by methoxycarbonylation with dimethyl carbonate and sodium hydride in dimethoxyethane at reflux temperature in 84 % yield. Although the cyclopropane ring cleavage of 1 or 5 with 99 % formic acid at 70-80°C afforded the corresponding formate (C₂₋₈ bond cleavage) and its isomer (C₁₋₂ bond cleavage)⁵⁾, in the case of 6 with methoxycarbonyl group at C-4 position, the only C₂₋₈ bond was selectively cleaved with 99 % formic acid in the presence of conc. sulfuric acid at room temperature to the desired formate and a subsequent treatment with sodium methoxide afforded the hydroxyl keto ester 7 in 88 % yield from 6. Deoxygenation of ketone at C-3 position was achieved by the thioketalization followed by reduction with Raney Ni

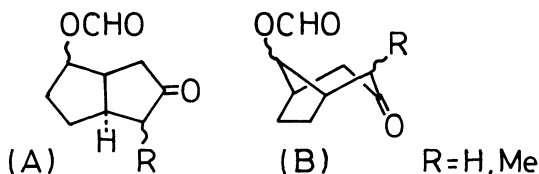


(W-2). Thus, $\underline{7}$ gave the hydroxy ester $\underline{8}$ in 80 % yield from $\underline{7}$. $\underline{8}$ was oxidized with Jones reagent to give the keto ester $\underline{9}$ which was converted into the corresponding cyanohydrin by a treatment with KCN and AcOH in EtOH at 30°C in good yield. This was dehydrated with phosphoryl chloride in pyridine to give αβ-unsaturated nitrile $\underline{10}$ in 84 % yield. $\underline{10}$ was hydrolyzed with potassium hydroxide in ethylene glycol at 160-180°C to its dicarboxylic acid which was treated with diazomethane to give the corresponding diester $\underline{11}$ in 79 % yield. $\underline{11}$ was deprotonated with lithium diisopropyl amide-hexamethylphosphoramide complex in THF at -78°C to produce the lithium enolate of $\underline{11}$ in situ, which was quenched with acetic acid to give the βγ-unsaturated ester $\underline{12}$ in 84 % yield (ratio: αβ/βγ = 1.0/10.4 by pmr). Ozonolysis of the mixture of $\underline{12}$ followed by reductive workup with Zn/AcOH directly led to (±)-forsythide aglycon dimethyl ester ($\underline{4b}$) after purification by preparative TLC as an oil, which was epimeric mixture at C-1 position in 66 % yield (ratio: α-OH/β-OH = 3.5/1.0 by pmr).

The pmr and ir spectra of synthetic $\underline{4b}$ were consistent with those of the compound⁶⁾ derived from natural forsythide. As described above, the stereo-controlled and facile synthesis of (±)-forsythide aglycon dimethyl ester ($\underline{4b}$) was achieved starting from versatile synthon $\underline{6}$.

References

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5. The ratio of C₂₋₈ bond cleavage (A) and C₁₋₂ bond cleavage (B) was 4 : 1 in either case of $\underline{1}$ and $\underline{5}$.



6. We thank Prof. T. Miwa and Dr. K. Furuichi for kindly providing copies of the pmr and ir spectrum of (±)-forsythide aglycon dimethyl ester.

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